Synthesis and Screening of New Chiral Ligands for the Copper-Catalysed **Enantioselective Allylic Substitution**

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This paper is dedicated to Professor Dieter Seebach, on the occasion of his 65th birthday, for his prominent contributions to synthetic chemistry

The synthesis of the new chiral ligands **6ae**, **8ae**, **9ae**, and **11ae** starting from the chiral β -[(Boc)amino]sulfonamide **3ae** is reported. The β -amino group of **3ae** was deprotected and condensed with 3,5dichlorosalicylaldehyde (4) to yield the known Schiff base 5ae, which was then reduced to the amino compound 6ae (Scheme 3). Alternatively, condensation of the free amino compound with 2-(diphenylphosphanyl)benzaldehyde (7) afforded the imino ligand **8ae** which upon reduction yielded the amino ligand **9ae** (Scheme 4). The free amino compound derived from 3ae was also coupled with 2-(diphenylphosphanyl)benzoic acid (10) to give ligand **11ae** (Scheme 5). These ligands were tested in the copper-catalysed allylic substitution reaction of cinnamyl (= 3-phenylprop-2-enyl) phosphate 12 with diethylzinc as a nucleophile. Ligands 5ae, 6ae, 8ae, and **11ae** gave excellent ratios (100:0) of the $S_N 2'/S_N 2$ products (*Scheme 6* and *Table 1*). Ligand **11ce**, identified from the screening of a small library of ligands of general formula 11, promoted the allylic substitution reaction with moderate enantioselectivity (40% for the $S_N 2'$ product 13 (Scheme 8 and Table 3)).

Introduction. - The efficiency (activity and selectivity) of a ligand for asymmetric catalysis depends on a subtle balance of the electronic, geometric, and steric influences between the ligand, the metal center, and the substrate. In such a complicated scenario, the development of effective new ligands and metal complexes based purely on intuition and trial-and-error is a very challenging task. The use of combinatorial methodologies for the rapid synthesis and screening of a large number of structures represents an important breakthrough in this area¹). Two different basic approaches have been considered: optimization of the reaction conditions (solvent, temperature, stoichiometry, different ligands or metal ions) and the synthesis of new ligands via a modular building-block strategy, where the stereoelectronic properties of a metalbinding site (e.g. a phosphine, a sulfonamide, or a Schiff base) are tuned by variation of the substituents and side chains. In the case of screening members of a library containing ligands for enantioselective catalysis, the identification of a hit requires a demanding selection procedure, since the screening ultimately involves catalysis of a

¹⁾ For reviews on the combinatorial development of new catalysts, see [1]. See also [2a - k]. For the screening of a pooled stereoselective polymerization catalyst library, see [21].

reaction and analysis of its stereochemical outcome. For this reason, a combinatorial system that allows the synthesis of discrete isolated compounds is usually chosen. Parallel synthesis (as opposed to the 'split and pool' methodology) allows one to know the identity of each ligand and keeps the ligands separate so that screening of individual complexes can be performed.

We have recently developed a new family of chiral *Schiff*-base ligands of general structure **5** (*Scheme 1*), which contains a set of different metal-binding sites (a phenol, an imine, and a secondary sulfonamide moiety), with the expectation that such a multidentate array would favor the formation of organometallic complexes with well-organized spatial arrangements, and with the goal of obtaining ligands for asymmetric catalysis capable of broad applicability. The main feature of these ligands is their modular assembly through the subsequent coupling of the three components (*Scheme 1*), namely sulfonyl chlorides **1**, amines **2**, and aldehydes **4**, which make these ligands well-suited for a combinatorial development. A library of ligands **5** (125 compounds) was synthesized in solution (with resin scavenging of excess reagents and by-products) and tested in the copper-catalysed, conjugate addition of Et₂Zn to enones [3a,c] and nitroolefins [3b].





Results and Discussion. – To broaden the scope of this approach, we undertook a variation of the ligand structure by transformation or substitution of the phenolic *Schiff*-base moiety. The newly synthesized ligands **6**, **8**, **9**, and **11** (*Fig.*) containing the chiral β -aminosulfonamide substructure were screened in the copper-catalysed allylic substitution reaction of cinnamyl phosphate with diethylzinc as a nucleophile.

A small library of ligands of general formula **11** were prepared and screened for enantioselective allylic substitution.

Synthesis of the Ligands. The synthesis of a few representatives of the different ligand classes 6, 8, 9, and 11 was undertaken starting from sulfonyl chloride 1a ($R^1 = PhCH_2$) [4]. Sulfonyl chloride 1a was coupled to an excess of amine 2e (1.2 equiv.) in the presence of dimethylketene methyl trimethylsilyl acetal (DMTA) (2.0 equiv.) and a catalytic amount (0.2 equiv.) of *N*,*N*-dimethylpyridin-4-amine (DMAP) in CH₂Cl₂ (*Scheme 2*) [4a]. Once the coupling was complete, the reaction mixture was washed



with a saturated citric acid solution, and sulfonamide **3ae** was obtained in 88% yield without the need for any further purification.

Removal of the Boc protecting group of **3ae** (25% CF₃COOH in CH₂Cl₂) afforded the trifluoroacetate ammonium salt, which was then liberated (NaHCO₃). The resulting amino compound was then transformed into the different ligands **5ae**, **6ae**, **8ae**, **9ae**, and **11ae**. Thus, condensation with 3,5-dichlorosalicylaldehyde (4) in MeOH yielded the *Schiff*-base ligand **5ae** in practically quantitative yield without the need for any further purification. *Schiff*-base ligand **5ae** was then reduced to the amino compound **6ae** by reaction with NaBH₃CN in MeOH in the presence of HCl [5].

Formation of the Schiff-base ligand **8ae** required a different synthetic protocol: condensation of the amino compound obtained by deprotection of β -[(Boc)amino]sulfonamide **3ae** with 2-(diphenylphosphanyl)benzaldehyde was performed in dry toluene at room temperature with MgSO₄ as dehydrating agent [6]. Removal of the salt by filtration and evaporation of toluene afforded **8ae**, which was not further purified. Reduction of the imino moiety of ligand **8ae** under the same conditions as described above yielded the amino ligand **9ae** in 63% yield.

For the synthesis of **11ae**, the chiral β -aminosulfonamide derived from **3ae** was coupled with 2-(diphenylphosphanyl)benzoic acid (**10**) in the presence of BOP ((1*H*-benzotriazol-1-yloxy)tris(dimethylamino)phosphonium hexafluorophosphate) and Et₃N [7] (70% yield). Use of EDC (*N*-[3-(dimethylamino)propyl]-*N*'-ethylcarbodiimide]) and DMAP in THF [8] proved less satisfactory (yield $\leq 29\%$).

With these ligands in our hands, we considered their application in the coppercatalysed asymmetric allylic substitution.

Allylic-Substitution Reaction. The substitution of allylic alcohol derivatives by hard C-nucleophiles, such as organometallics, has recently attracted much attention, and

Scheme 3. Synthesis of Schiff-Base Ligand 5ae and Reduction to the Amino Ligand 6ae



Scheme 4. Synthesis of Imino Ligand 8ae and Reduction to the Amino Ligand 9ae



several methods have been developed for the control of the regio- and stereochemistry in this type of reaction. The leaving group can be displaced either in an α (S_N2) or γ (S_N2') fashion by the organometallic reagent depending on the substrate, leaving group, organometallic, and metal source. Some enantioselective protocols that use catalytic amounts of a chiral copper complex, and organozinc [9] or *Grignard* [10] reagents have recently appeared: high ee were obtained in special cases with hindered organozinc reagents [9a] or 3-substituted cinnamyl (= 3-phenylprop-2-enyl) derivatives leading to



quaternary C-atoms [9b]. We decided to test the chiral Schiff-base ligand **5ae** and the modified ligands 6ae, 8ae, 9ae, and 11ae in the copper-catalysed allylic substitution of cinnamyl derivatives with diethylzinc. At the beginning, we screened several cinnamyl derivatives to find a suitable substrate; it was soon clear that halogenated derivatives (chloride and bromide) did not give a regioselective substitution reaction (low $S_N 2'/S_N 2$ ratio) either in the presence or absence of ligands. In the case of cinnamyl chloride (= 3chloro-1-phenylprop-1-ene), a smooth reaction took place in toluene at -20° , even in the absence of any copper source, yielding an equimolar mixture of $S_N 2'$ and $S_N 2$ products 13 and 14²). Following the discovery by *Hoveyda et al.* of cinnamyl phosphate 12 as a suitable substrate for the enantioselective copper-catalysed allylic substitution with alkylzinc reagents [9b], we screened ligands 5ae, 6ae, 8ae, 9ae, and 11ae using CuCN as the metal salt (see Scheme 6 and Table 1). All ligands (except 9ae, Entry 7) gave fair-to-good yields in THF (the reaction did not proceed in toluene) with practically complete $S_N 2'$ regioselectivity (13/14 100:0). As for the enantioselectivity, all the ligands tested gave racemic mixtures, except for 11ae which gave 13 with a very modest 9% ee (Entry 8).

Scheme 6. Copper-Catalysed Allylic Substitution of Cinnamyl Phosphate 12 with Et₂Zn



Entry	L*	CuY	Solvent	$T [^{\circ}]$	Conversion ^a)	$S_{\rm N}2'/S_{\rm N}2~(13/14)^{\rm a})$	ee (13) ^a)
1	_	_	THF	- 30	0	-	_
2	-	CuCN	THF	- 30	0	_	_
3	_	CuCN	THF	0	39	100:0	_
4	5ae	CuCN	THF	- 30	100	100:0	0
5	6ae	CuCN	THF	- 30	55	100:0	0
6	8ae	CuCN	THF	-30	53	100:0	0
7	9ae	CuCN	THF	- 30	0	_	_
8	11ae	CuCN	THF	-30	91	100:0	9

Table 1. Copper-Catalysed Allylic Substitution of Cinnamyl Phosphate 12 with Diethylzinc (see Scheme 6)

²) *Feringa et al.* showed that cinnamyl halides can be used as substrates for the enantioselective coppercatalysed allylic substitution only in strongly coordinating solvent (*e.g.*, diglyme) [9c]. A screening of the copper sources was then performed with **11ae** as ligand (*Table 2*), and a more-interesting 30% ee was obtained by lowering the temperature to -55° and running the reaction in the presence of copper(I) triflate dimer (benzene complex) (*Entry 1*). The use of copper(II) triflate gave almost complete S_N2' regioselectivity and the same ee, albeit in much lower yield (*Entry 2*).

 Table 2. Copper-Catalysed Allylic Substitution of Cinnamyl Phosphate 12 with Diethylzinc, Chiral Ligand 11ae, and Different Copper Sources

Entry	L*	CuY	Solvent	$T\left[\circ ight]$	Conversion ^a)	$S_{\rm N}2'/S_{\rm N}2~(13/14)^{\rm a})$	ee (13) ^a)
1	11ae	$[(CuOTf)_2 \cdot C_6H_6]$	THF	- 55	95	89:11	30
2	11ae	$Cu(OTf)_2$	THF	- 55	15	100:0	30
3	11ae	CuCN	THF	- 55	0	-	-
4	11ae	$CuBr \cdot SMe_2$	THF	- 55	10	85:15	0
5	11ae	CuI	THF	- 55	0	-	-

Based on these results, we undertook the synthesis of a small library of ligands of general formula **11**, changing the residues at the β -position of the β -aminosulfonamide (R¹) and at the N-atom of the amide moiety (R²). Thus 14 different ligands were prepared starting from the 4-different β -aminosulfonyl chlorides [4] **1a**-**d** (R¹=Me, PhCH₂, ⁱPr, ⁱBu) and 7 different amines **2e**-**k** according to the general procedure described above³)⁴) (*Scheme* 7).)

The screening of the ligands required complexation *in situ* with $[(CuOTf)_2 \cdot C_6H_6]$ in THF at room temperature for 45 min. After addition of phosphate **12**, the reaction mixture was cooled to -78° , and Et₂Zn was added. The mixtures were warmed to -55° and stirred for 20 h before quenching with NH₄Cl (*Scheme 8*). The ee and conversions were measured by GC with a chiral column and with decane as an internal standard. The absolute configuration of the stereogenic center of **13** was assigned as (*S*), based on the sign of the optical rotation [11]. The results are summarized in *Table 3*.

From these results, one can draw the following conclusions: *a*) The stereogenic center bearing R¹ dictates the absolute configuration of compound **13**, and not the stereogenic center present in R² (*cf. Entries 4* and 5, *Table 3*). b) The ee of **13** depends on a subtle balance of the mutual influences of R¹ and R²; no linear relationship between the steric hindrance of these residues and ee can be clearly identified. c) The best combination of R¹ and R² (R¹ = ⁱPr and R² = (S)-ChxCH(Me)), *i.e.* ligand **11ce**, gave **13** with an ee of 40% and a ratio **13/14** of 90:10.

Conclusions. – We reported the synthesis of the new chiral ligands 6, 8, 9, and 11 (*Fig.*), encompassing the chiral β -aminosulfonamide substructure. These ligands were

³) We did not synthesize the full matrix of all possible different ligands (*i.e.* 7 · 4 = 28), but only half of them (14). We followed a sort of 'positional scanning' approach, *e.g.*, synthesized ligands **11ce** - **ck** with all seven possible R² (**e** - **k**) in the case of the best R¹ (**c**).

⁴) In some cases, after removal of the Boc protecting group from the amino group, the amino derivative was obtained from the intermediate trifluoroacetate salt by treatment with Et₃N (2 equiv.) in THF. The resulting solution was used directly in the following coupling step with 2-(diphenylphosphanyl)benzoic acid.

Scheme 7. Synthesis of the Library of Ligands 11



Scheme 8. Screening of the Library of Ligands 11 in the Copper-Catalysed Allylic Substitution of Cinnamyl Phosphate 12 with Et_2Zn



tested in the copper-catalysed allylic substitution reaction of cinnamyl phosphate with diethylzinc as a nucleophile. Ligands **5**, **6**, **8**, and **11** gave excellent ratios (100:0) of the $S_N 2'/S_N 2$ products **13** and **14**. The screening of a small library of ligands of general formula **11** (14 compounds) allowed the identification of **11ce** as the best ligand for enantioselective allylic substitution (ee 40% for the $S_N 2'$ product **13**). Work is now in progress to test the applicability of these ligand structures to other catalytic asymmetric reactions of interest.

Entry	L*	L^* R^1 R^2		Conversion ^a)	$S_{\rm N}2'/S_{\rm N}2~(13/14)^{\rm a})$	ee (13) ^a) ^b)
1	11ae	PhCH ₂	(S)-ChxCH(Me)	95	89:11	30
2	11be	Ме	(S)-ChxCH(Me)	50	89:11	30
3	11de	ⁱ Bu	(S)-ChxCH(Me)	76	89:11	38
4	11ce	ⁱ Pr	(S)-ChxCH(Me)	93	90:10	40
5	11cf	ⁱ Pr	(R)-ChxCH(Me)	55	87:13	21
6	11ci	ⁱ Pr	Ph ₂ CH	41	83:17	13
7	11cg	ⁱ Pr	PhCH ₂	75	57:43	0
8	11ch	ⁱ Pr	ⁱ Pr	7	75:25	0
9	11cj	ⁱ Pr	(S)-CH ₂ (OH)CH(Ph)	80	80:20	13
10	11ck	ⁱ Pr	(S)-PhCH(Me)	3	-	_
11	11ak	PhCH ₂	(S)-PhCH(Me)	50	90:10	27
12	11dk	ⁱ Bu	(S)-PhCH(Me)	17	90:10	0
13	11ah	$PhCH_2$	ⁱ Pr	65	90:10	22
14	11bg	Me	PhCH ₂	100	70:30	13

Table 3. Copper-Catalysed Allylic Substitution of Cinnamyl Phosphate **12** with Diethylzinc, $[(CuOTf)_2 \cdot C_6H_6]$, and a Library of Ligands **11** (see Scheme 8)

^a) Determined by chiral GC analysis (*Megadex DACTBSβ*) with decane as internal standard. ^b) The absolute configuration of the stereogenic center of **13** was assigned as (S), based on the sign of the optical rotation [11].

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Experimental Part

General. Manipulations involving air-sensitive compounds were carried out under Ar by Schlenk and syringe techniques. Anh. solvents were obtained from Na (toluene), sodium benzophenone ketyl (oxidodiphenylmethyl sodium salt; THF and Et₂O), or by refluxing over CaH₂ for at least 4 h prior to use. Reagents were used as received without any further purification and were generally purchased from Aldrich and Fluka AG. Compounds **3ae** [3c], **5ae** [3c], and **12** [9a] were obtained as previously described. Anal. TLC: Merck silica-gel 60 F_{254} glass plates; visualization with UV light and by staining with a cerium reagent followed by heating. Flash chromatography [12] (FC): silica gel 60 (230–400 mesh) from Macherey Nagel. GC: Dani GC-3800 instrument equipped with a FID and a chiral capillary column (Megadex DACTBS β , decane as internal standard); $t_{\rm R}$ in min. Optical rotations: Perkin-Elmer 241 polarimeter. IR Spectra: Perkin-Elmer 681; in cm⁻¹. NMR Spectra: Bruker instruments (AC-200 and 400 Avance); δ in ppm rel. to SiMe₄; J in Hz.

1. (2S)-N-[(1S)-1-Cyclohexylethyl]-2-[(3,5-dichloro-2-hydroxybenzyl)amino]-3-phenylpropane-1-sulfonamide (6ae). To a soln. of 5ae (240 mg, 0.48 mmol) in MeOH (27 ml), NaBH₃CN was added (155 mg, 2.46 mmol). The soln. was stirred for 5 min and 37% (*w*/*w*) conc. HCl soln. (263 µl) was added dropwise. The yellow soln. discolored, and stirring was maintained for 2 h at r.t. A 3m KOH soln. was added until the pH of the soln. became basic, the mixture was diluted with H₂O (50 ml) and extracted several times with CH₂Cl₂. The combined org. extract was washed with brine, dried (Na₂SO₄), and evaporated; and the residue was purified by FC (hexane/AcOEt 80:20): 171 mg (71%) of pure 6ae. ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.75–0.83 (*m*, 14 H, Chx, Me); 3.12–2.83 (*m*, CH₂SO₂, ChxCH(Me), PhCH₂); 3.49–3.45 (*m*, ChxCH(Me)); 3.99 (*s*, CH₂NH); 4.68 (br. *s*, NHSO₂); 7.36–6.87 (*m*, arom. 7 H). ¹³C-NMR (CDCl₃, 100.13 MHz): 19.4; 26.0; 26.2; 28.6; 28.8; 39.4; 43.5; 49.5; 54.9; 55.2; 55.8; 121.7; 123.6; 124.9; 126.7; 127.2; 128.6; 129.0; 129.3; 136.5; 152.3. IR (CHCl₃): 1009 (CN), 1333 (SO₂), 1435, 2857, 2932, 3387 (NH). [*a*]_D²⁵ = +15.2 (*c* = 0.25, CHCl₃). HR-EI-MS: 498.1276 (C₂₄H₃₂Cl₂N₂O₃S⁺; calc. 498.15).

2. (2S)-N-[(1S)-1-Cyclohexylethyl]-2-{[2-(diphenylphosphanyl)benzylidene]amino]-3-phenylpropane-1sulfonamide (8ae). β -[(Boc)amino]sulfonamide 3ae (50 mg, 0.12 mmol) was dissolved in 25% CF₃COOH in CH₂Cl₂ (5.2 ml). The resulting soln. was stirred for 1 h, and sat. aq. NaHCO₃ soln. was added until pH 7. The aq. phase was extracted with AcOEt (6 × 10 ml) and the combined org. extract dried (Na₂SO₄) and evaporated. The residue was suspended in dry toluene (5 ml), and 2-(diphenylphosphanyl)benzaldehyde (33 mg, 0.12 mmol) and anh. MgSO₄ (14 mg, 0.12 mmol) were added. The resulting suspension was stirred at r.t. for 15 h under Ar. The solid salts were filtered and washed with AcOEt, and the combined filtrates were evaporated to yield 70 mg (quant.) of **8ae**. ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.81–0.74 (*m*, 14 H, Chx, Me); 2.87–2.57 (*m*, CH₂SO₂); 3.29–3.19 (*m*, ChxCH(Me), PhCH₂); 3.95–3.92 (*m*, CHN=CH); 4.13 (*d*, *J* = 8.78, NHSO₂); 7.38–6.93 (*m*, 19 arom. H); 8.64 (*s*, CH=N). ¹³C-NMR (CDCl₃, 100.13 MHz): 18.8; 26.1; 26.3; 28.5; 29.0; 42.3; 43.7; 54.5; 57.8; 67.9; 126.6; 128.3; 128.6; 128.8; 128.9; 129.4; 129.6; 130.5; 133.9; (*d*, *J* = 19.6); 135.0 (*d*, *J* = 18.8); 136.9 (*d*, *J* = 19.5); 137 (*d*, *J* = 19.7); 141.8 (*d*, *J* = 27.8); 161.2; 161.4. IR (CHCl₃): 960, 1340 (SO₂), 1456 (CH₂, Me), 1640 (C=N), 3412 (NH). [*a*]_D² = -155.71 (*c* = 0.21, CHCl₃). HR-EI-MS: 596.2664 (C₃₀H₄₁N₂O₂PS⁺; calc. 596.2626).

3. (2S)-N-[(1S)-1-Cyclohexylethyl]-2-[[2-diphenylphosphanyl)benzyl]amino]-3-phenylpropane-1-sulfonamide (9ae). As described for 6ae, with 8ae (290 mg, 0.48 mmol), MeOH (27 ml), NaBH₃CN (149 mg, 2.37 mmol), and 37% (*w*/*w*) conc. HCl soln. (200 µl): 114 mg (39%) of pure 9ae. ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.84–0.82 (*m*, 15 H, Chx, Me, NH); 2.92–2.61 (*m*, CH₂SO₂); 3.07–2.99 (*m*, PhCH₂); 3.27–3.25 (*m*, Chx, CHMe); 3.51–3.47 (*m*, PhCH₂CH); 4.26–3.93 (*dd*, J = 114.69, 12.89, CH₂NH); 4.97 (*d*, J = 8.38, NHSO₂); 7.49–6.91 (*m*, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 18.7; 26.1; 26.3; 28.5; 28.7; 40.0; 43.7; 49.7; 49.9; 54.3; 55.8; 56.6; 71.8; 126.6; 127.6; 128.6; 128.8; 128.9; 129.2; 129.4; 129.5; 133.7 (*d*, J = 18.5); 134.0; 135.7 (*d*, J = 18.8); 136.4 (*d*, J = 19.5); 137.3 (*d*, J = 19.7); 143.7 (*d*, J = 23.6). IR (nujol): 710 (C–P), 746 (S–N), 1099 (CN), 1310 (SO₂), 3283 (NH). [a] $\frac{15}{25}$ = +5.9 (*c* = 0.51, CHCl₃). HR-EI-MS: 598.2754 (C₃₆H₄₃N₂O₂PS⁺; calc. 598.2783).

4. Ligands **11.** General Procedure a. β -(Boc)aminosulfonamide **3** (1 equiv.) was dissolved in 25% CF₃COOH in CH₂Cl₂ (0.1m based on the sulfonamide). The resulting soln. was stirred for 1 h, and sat. aq. NaHCO₃ soln. was added until pH 7. The aq. phase was extracted with AcOEt (6×10 ml). The combined org. extracts were dried (Na₂SO₄) and evaporated. In a separate flask, (1*H*-benzotriazol-1-yloxy)tris(dimethylamino)phosphonium hexafluorophosphate (BOP, 1 equiv.) was added to a soln. of 2-(diphenylphosphanyl)benzoic acid in dry THF (1 equiv., 0.07m) and Et₃N (1 equiv.). After 10 min, a soln. of the deprotected amino compound in THF (0.35m) was added, and the mixture was stirred overnight. The mixture was evaporated and AcOEt (10 ml) was added. The soln. was washed with 5% HCl, sat. NaHCO₃ soln., and brine, and was dried (Na₂SO₄). Purification by FC (hexane/AcOEt:) afforded pure **11**.

General Procedure b. β -[(Boc)amino]sulfonamide **3** (1 equiv.) was dissolved in 25% CF₃COOH in CH₂Cl₂ (0.1M based on the sulfonamide). The resulting soln. was stirred for 1 h and then evaporated. The residue was dissolved in THF (0.35M based on the sulfonamide), and Et₃N (2 equiv.) was added. The resulting soln. was stirred for 30 min and directly used in the subsequent coupling step. In a separate flask, BOP (1 equiv.) was added to a soln. of 2-(diphenylphosphanyl)benzoic acid in dry THF (1 equiv., 0.07M) and Et₃N (1 equiv.). After 10 min, the soln. of the deprotected amino compound in THF/CF₃COOH was added, and the mixture was stirred overnight. After evaporation, AcOEt (10 ml) was added. The soln. was washed with 5% HCl soln. sat. NaHCO₃ soln., and brine, and was dried (Na₂SO₄). Purification by FC (hexane/AcOEt 8:2) afforded pure **11**.

N-f(1S)-1-Benzyl-2-f[(1S)-1-cyclohexylethyl]sulfamoyl]ethyl]-2-(diphenylphosphanyl)benzamide (11ae): ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.75–0.89 (m, 14 H, Chx, Me); 3.03–2.84 (m, CH₂SO₂); 3.14–3.13 (m, PhCH₂); 3.31–3.26 (m, ChxCH(Me)); 4.73–4.68 (m, PhCH₂CH); 4.79 (d, J=8.68, NHSO₂); 6.5 (br., NHCO); 7.38–7.02 (m, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 19.4; 26.1; 26.3; 28.5; 29.0; 39.6; 43.6; 48.0; 54.8; 55.3; 57.0; 126.9; 127.6; 128.7; 129.1; 129.4; 130.4; 133.7 (d, J=19.1); 134.3; 136.7. IR (nujol): 734 (C–P), 750 (S–N), 1322 (SO₂), 1646 (C=N), 3440 (NH). $[a]_D^{25} = +9.0$ (c = 0.40, CHCl₃). HR-EI-MS: 612.2587 (C₃₆H₄₁N₂O₃PS⁺; calc. 612.2576).

N-f(IS)-2-f[(IS)-1-Cyclohexylethyl]sulfamoyl]-1-methylethyl]-2-(diphenylphosphanyl)benzamide (11be): ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.79–0.96 (*m*, 17 H, Chx, 2 Me); 3.23–3.08 (*m*, CH₂SO₂); 3.38–3.32 (*m*, ChxCH(Me)); 4.57–4.51 (*m*, NHCH(Me)); 4.69 (*d*, J = 8.74, NHSO₂); 7.67–6.96 (*m*, 14 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 19.2; 19.7; 26.1; 26.3; 28.5; 29.0; 42.5; 43.7; 54.7; 58.1; 127.8; 128.6; 129.0; 130.3; 133.8 (*d*, J = 16.5); 134.1; 134.8 (*d*, J = 18.0); 136.3 (*d*, J = 19.5); 136.8 (*d*, J = 19.5); 142.8 (*d*, J = 26.8); 168.7. IR (nujol): 698 (C–P), 743 (S–N), 1130, 1306 (SO₂), 1520, 1640 (C=N), 3221 (NH). [a]²⁵_D = +14.0 (c = 0.30, CHCl₃). HR-EI-MS: 536.2216 ($C_{30}H_{37}N_2O_3PS^+$; calc. 536.2262).

N-{(1S)-1-{[[(1S)-1-Cyclohexylethyl]sulfamoyl]methyl]-3-methylbutyl]-2-(diphenylphosphanyl)benzamide (11de): ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.87–2.06 (m, 23 H, 3 Me, Chx, Me₂CH, Me₂CHCH₂); 3.16– 3.14 (m, CH₂SO₂); 3.38 (m, ChxCH(Me)); 4.44 (br. m, ¹BuCH); 4.80–4.78 (m, NHSO₂); 6.38 (br. s, CONH); 7.40–7.26 (m, 14 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 19.4; 21.8; 22.8; 24.7; 26.2; 26.3; 28.6; 29.0; 42.9; 43.7; 44.8; 54.8; 57.4; 128.6; 130.3; 133.7 (d, J = 19.2); 134.3; 135.0 (d, J = 18.8); 136.9 (d, J = 19.5); 137.0 (d, J = 19.7); 141.8 (d, J = 27.8); 169.6. IR (nujol): 720 (C–P), 753 (S–N); 1168, 1315 (SO₂), 1512, 1651 (C=N), 3378 (NH). [α]₂₅²⁵ = +16.5 (c = 0.51, CHCl₃). HR-EI-MS: 578.2705 (C₃₃H₄₃N₂O₃PS⁺; calc. 578.2732).

$$\begin{split} & \text{N-}f(1\text{S})\text{-}1\text{-}f\{[(1\text{S})\text{-}1\text{-}Cyclohexylethyl]sulfamoyl]\text{methyl]}\text{-}2\text{-}methylpropyl]\text{-}2\text{-}(diphenylphosphanyl)\text{benza-mide} (11\text{ce}): ^{1}\text{H-NMR} (\text{CDCl}_3, 400 \text{ MHz}, 20^\circ): 0.87 (d, J = 6.8, \text{Me}); 0.91 (d, J = 6.7, \text{Me}); 1.82 - 0.95 (m, 14 \text{ H}, \text{Chx}, \text{ Me}); 2.04 - 1.96 (m, \text{Me}_2\text{C}H); 3.21 - 3.10 (m, \text{CH}_2\text{SO}_2); 3.46 - 3.37 (m, \text{ChxCH}(\text{Me})); 4.52 - 4.46 (m, ^{1}\text{PrC}H); 5.01 (d, J = 8.6, \text{NHSO}_2); 6.26 (d, J = 9.1, \text{NHCO}); 7.71 - 6.98 (m, 14 \text{ arom}. \text{ H}). ^{13}\text{C-NMR} (\text{CDCl}_3, 100.13 \text{ MHz}): 18.2; 18.8; 19.3; 26.1; 26.2; 26.3; 28.5; 29.0; 31.3; 43.7; 51.3; 54.7; 54.9; 127.6; 127.7; 128.5; 128.6; 128.7; 128.8; 129.0; 130.2; 133.6 (d, J = 19.9); 134.4; 135.0 (d, J = 18.8); 136.9 (d, J = 19.5); 137.0 (d, J = 19.7); 141.8 (d, J = 27.8); 169.6 \text{ IR} (\text{nujol}): 723 (\text{C}-\text{P}), 746 (\text{S}-\text{N}), 1161, 1310 (\text{SO}_2), 1512, 1651 (\text{C}=\text{N}), 3430 (\text{NH}). \\ \hline [\alpha]_{25}^{25} + 40.6 (c = 0.50, \text{CHC}_3). \text{ HR-EI-MS: 564.2601 (C}_{32}\text{H}_4\text{N}_2\text{O}_3\text{PS}^+; \text{calc}. 564.2575). \end{split}$$

$$\begin{split} & \text{N-}\{(1\text{S})\text{-}1\text{-}\{[1(\text{R})\text{-}1\text{-}Cyclohexylethyl]\text{sulfamoyl}\text{methyl}\text{-}2\text{-}methylpropyl}\text{-}2\text{-}(diphenylphosphanyl)\text{benz-amide} (11\text{cf}): ^{1}\text{H-NMR} (\text{CDCl}_3, 400 \text{ MHz}, 20^\circ): 0.90 - 0.88 (m, 2 \text{ Me}); 1.94 - 0.95 (m, 15 \text{ H}, \text{Chx}, 1 \text{ Me}, \text{Me}_2\text{C}H); \\ & 3.18 - 3.16 (m, \text{CH}_2\text{SO}_2); 3.42 - 3.37 (m, \text{ChxCH}(\text{Me})); 4.52 - 4.48 (m, ^{1}\text{PrC}H); 5.37 (m, \text{NHSO}_2); 6.28 (br., \text{NHCO}); 7.72 - 7.03 (m, 14 \text{ arom}. \text{H}). ^{13}\text{C-NMR} (\text{CDCl}_3, 100.13 \text{ MHz}): 18.4; 18.6; 19.0; 26.1; 26.2; 28.7; 31.7; \\ & 43.8; 51.1; 54.2; 54.6; 127.6; 128.4; 128.5; 128.6; 128.7; 130.2; 133.5 (d, J = 19.5); 133.7; 134.5 (d, J = 18.8); 136.9 (d, J = 19.5); 137.0 (d, J = 19.7); 142.0 (d, J = 27.8); 169.6 \text{ IR} (\text{nujol}): 710 (\text{C}-\text{P}), 750 (\text{S}-\text{N}), 1130, 1310 (\text{SO}_2), \\ & 1521, 1645 (\text{C}=\text{N}), 3280 (\text{NH}). [a]_{\text{D}}^{25} = +55.4 (c = 0.52, \text{CHCl}_3). \text{ HR-EI-MS: } 564.2591 (\text{C}_{32}\text{H}_{41}\text{N}_2\text{O}_3\text{PS}^+; \text{ calc.} \\ & 564.2576). \end{split}$$

N-f(1S)-I-[(Benzhydrylsulfamoyl)methyl]-2-methylpropyl]-2-(diphenylphosphanyl)benzamide (11ci). ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.62–0.57 (m, 2 Me); 1.45–1.40 (m, Me₂CH); 3.00–2.69 (m, PhCH₂); 4.12–4.05 (m, Me₂CHCH); 5.8 (d, J = 8.96, NHSO₂); 6.07 (br. s, NHCO); 7.50–6.90 (m, 24 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 17.6; 18.4; 31.8; 50.3; 54.5; 61.4; 127.2; 127.7; 128.0; 128.2; 128.4; 128.9; 130.2; 133.5 (d, J = 19.6); 134.6; 135.0 (d, J = 18.8); 136.9 (d, J = 19.5); 137 (d, J = 19.7); 141.2 (d, J = 27.8); 169.6. IR (nujol): 723 (C–P), 750 (S–N), 1315 (SO₂), 1536, 1645 (C=N), 3452 (NH). [a]²⁵_D = 66.83 (c = 0.41, CHCl₃). HR-EI-MS: 620.2273 ($C_{37}H_{37}N_2O_3PS^+$; calc. 620.2263).

N-f(1S)-1-f(Benzylsulfamoyl)methyl]-2-methylpropyl]-2-(diphenylphosphanyl)benzamide (11cg): ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.74–0.72 (d, J = 6.84, Me); 0.78–0.76 (d, J = 7.53, Me); 1.68–1.80 (m, Me_2CH); 3.13–2.85 (m, CH_2SO_2); 4.43–4.25 (m, PrCH, PhCH₂); 6.30–6.08 (br., NHSO₂, NHCO); 7.85–7.07 (m, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 17.7; 18.0; 18.7; 31.1; 31.7; 47.2; 50.8; 51.4; 52.3; 54.0; 127.8; 128.3; 128.6; 128.7; 128.8; 129.2; 130.3; 131.8; 132.3; 133.5; 133.6 (d, J = 12.6); 133.8; 134.4; 137.1; 137.2; 170.0. IR (nujol): 696 (C–P), 723 (S–N), 1150, 1308 (SO₂), 1541, 1641 (C=N), 3190, 3356 (NH). [a]²⁵ = +48.04 (c =0.51, CHCl₃). HR-EI-MS: 544.1984 (C₃₁H₃₃N₂O₃PS⁺; calc. 544.1949).

2-(Diphenylphosphanyl)-N-{(1S)-1-[(isopropylsulfamoyl)methyl]-2-methylpropyl}benzamide (11ch): ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.86–0.89 (m, 2 Me); 1.26–1.24 (m, 2 Me); 1.57 (m, Me₂CH); 3.17–3.16 (m, CH₂SO₂); 3.68–3.63 (m, SO₂CH(Me)); 4.57–4.51 (m, ¹PrCH); 5.06–5.04 (m, NHSO₂); 6.20 (br., NHCO); 7.70–6.98 (m, 14 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 21.9; 22.7; 24.0; 24.4; 24.6; 43.0; 44.8; 46.3; 57.0; 127.9; 128.6; 128.9; 130.3; 133.6 (d, J = 19.9); 134.2; 135.0 (d, J = 18.8); 136.9 (d, J = 19.5); 137.0 (d, J = 19.7); 141.8 (d, J = 27.8); 169.6. IR (nujol): 698 (C–P), 735 (S–N), 900, 1009, 1159, 1300 (SO₂), 1539, 1649 (C=N), 2251, 3215, 3369 (NH). [a]²⁵ = +14.0 (c = 0.50, CHCl₃). EI-MS: 495, 433, 388, 348, 332, 304 (100), 289, 277, 261, 241, 226, 212, 199, 183, 165, 149, 139, 123, 108, 91, 83, 77, 69, 57, 43, 29.

2-(Diphenylphosphanyl)-N-{(1S)-1-{{[(1S)-2-hydroxy-1-phenylethyl]sulfamoyl]methyl]-2-methylpropyl]benzamide (11cj): ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.69–0.98 (m, 2 Me); 1.98–1.63 (m, Me₂CH); 3.83–2.65 (m, PhCH₂, CH₂OH, CH₂SO₂, CHCH₂OH); 4.82–4.58 (m, ¹PrCH); 6.45–6.12 (m, NHSO₂); 7.70–6.83 (m, 19 arom. H). IR (nujol): 730 (C–P), 748 (S–N), 1312 (SO₂), 1518, 1655 (C=N), 3398 (NH). $[a]_{25}^{25} = +124.7 (c = 0.51, CHCl₃)$. EI-MS: 573, 557, 513, 497, 423, 374, 363, 330, 305 (100), 289, 277, 254, 241, 226, 211, 199, 183, 165, 152, 120, 105, 91, 77, 69, 41.

2-(Diphenylphosphanyl)-N-{(1S)-2-methyl-1-{[[(1S)-1-phenylethyl]sulfamoyl]methyl]propyl]benzamide (**11ck**): ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.53 (d, J = 6.91, Me); 0.67 (d, J = 6.85, Me); 0.94–0.88 (m, Me₂CH); 1.58 (d, J = 6.99, Me); 2.87–2.59 (m, CH₂SO₂); 3.90–3.80 (m, Me₂CHCH); 4.63–4.53 (m, PhCH); 7.64–7.06 (m, 21 H, arom. H, NHSO₂, NHCO). ¹³C-NMR (CDCl₃, 100.13 MHz): 17.3; 17.9; 24.0; 30.8; 50.9; 52.0; 54.0; 126.8; 127.6; 128.5; 128.6; 128.7; 129.0; 129.1; 129.6; 129.7; 131.7; 131.8; 132.1; 133.6 (d, J = 11.9); 135.0 (d, J = 18.8); 136.9 (d, J = 19.5); 137 (d, J = 19.7); 141.8 (d, J = 27.8); 143.6; 169.2. IR (nujol): 723 (C–P), 754 (S–N), 1315 (SO₂), 1556, 1670 (C=N), 3267; 3384 (NH). [a]²⁵_D = + 132.0 (c = 0.50, CHCl₃). EI-MS: 531, 497, 481, 454, 391, 376, 348, 322, 305 (100), 292, 277, 258, 244, 227, 199, 183, 166, 152, 120, 105, 91, 77, 69.

N-f(1S)-1-Benzyl-2-[f(1S)-1-phenylethyl]sulfamoyl]ethyl]-2-(diphenylphosphanyl)benzamide (11ak): ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.53 (d, J = 7.03, Me); 2.48–2.80 (m, CH_2SO_2 , +PhCH₂); 4.68–4.30 (*m*, PhCH₂CH, PhCH); 5.76 (br. *d*, NHSO₂); 6.2 (br., NHCO); 7.71–6.80 (*m*, 24 arom. H). IR (nujol): 723 (C–P), 750 (S–N), 1161, 1312 (SO₂), 1510, 1648 (C=N), 3415 (NH). $[\alpha]_{D}^{25} = +60.2$ (*c* = 0.50, CHCl₃). EI-MS: 591, 529, 515, 501, 421, 409, 393, 357, 330, 305 (100), 289, 277, 254, 241, 226, 212, 199, 183, 165, 144, 117, 105, 91, 77, 65, 51.

2-(Diphenylphosphanyl)-N-{(IS)-3-methyl-1-{{[(IS)-1-phenylethyl]sulfamoyl]methyl]butyl]benzamide (**11dk**). ¹H-NMR (CDCl₃, 400 MHz, 20°): 0.62 (*d*, *J* = 6.5, 1 Me); 0.65 (*d*, *J* = 6.6, 1 Me); 1.07 – 1.48 (*m*, Me₂CH, Me₂CHCH₂); 1.57 (*d*, *J* = 5.1, 1 Me); 2.82 – 2.73 (*m*, CH₂SO₂); 4.10 – 4.01 (*m*, PhCH(Me)); 4.69 – 4.63 (*m*, ⁱBuCH); 7.63 – 7.06 (*m*, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 21.5; 22.5; 23.8; 23.9; 24.2; 43.1; 44.9; 53.9; 54.4; 126.5; 126.7; 127.5; 128.6; 128.7; 128.8; 128.9; 130.2; 132.1 (*d*, *J* = 17.3); 134.4; 135.0 (*d*, *J* = 18.8); 136.9 (*d*, *J* = 19.5); 137.0 (*d*, *J* = 19.7); 141.8 (*d*, *J* = 27.8). IR (nujol): 700 (C–P), 752 (S–N), 1159, 1308 (SO₂), 1528, 1645 (C=N), 3390 (NH). [*a*]_D²⁵ = +58.0 (*c* = 0.40, CHCl₃). HR-EI-MS: 573.2266 (C₃₃H₃₈N₂O₃PS⁺, [*M* + 1]⁺; calc. 573.2341).

N-[(1S)-1-Benzyl-2-(isopropylsulfamoyl)ethyl]-2-(diphenylphosphanyl)benzamide (11ah): ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.21 – 1.11 (m, 2 Me); 2.98 – 2.81 (m, CH₂SO₂); 3.12 (d, J = 6.27, PhCH₂); 3.59 – 3.51 (m, Me₂CH); 4.70 – 4.65 (m, PhCH₂CH); 4.97 – 4.95 (m, NHSO₂); 6.40 (br. s, NHCO); 7.41 – 6.98 (m, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 23.8; 24.3; 39.7; 46.3; 47.8; 55.0; 127.0; 127.5; 128.6; 128.7; 128.8; 129.4; 130.3; 133.7 (d, J = 19.9); 134.2; 135.2 (d, J = 19.0); 136.5 (d, J = 19.5); 137.2 (d, J = 19.7); 141.8 (d, J = 27.8); 169.6. IR (nujol): 733 (C–P), 748 (S–N), 1161, 1314 (SO₂), 1645 (C=N), 3227, 3356 (NH). [a]²⁵_D = +18.8 (c = 0.50, CHCl₃). HR-EI-MS 544.1954 (C₃₁H₃₃N₂O₃PS⁺; calc. 544.1949).

N-f(1S)-2-(Benzylsulfamoyl)-1-methylethyl]-2-(diphenylphosphanyl)benzamide (11bg): ¹H-NMR (CDCl₃, 400 MHz, 20°): 1.11 (d, J = 6.83, 1 Me); 1.90 (m, MeCH); 3.14 – 2.91 (dm, CH₂SO₂); 4.40 – 4.29 (m, PhCH₂, MeCH); 5.67 (t, J = 6.1, NHSO₂); 6.19 (d, J = 7.92, NHCO); 7.39 – 6.95 (m, 19 arom. H). ¹³C-NMR (CDCl₃, 100.13 MHz): 19.9; 42.3; 47.2; 57.1; 127.9; 128.2; 128.7; 129.0; 130.3; 133.8 (d, J = 19.9); 134.0; 135.4 (d, J = 18.8); 137.1 (d, J = 19.5); 137.2 (d, J = 19.7); 142.0 (d, J = 27.8). IR (nujol): 698 (C–P), 741 (S–N), 1150, 1317 (SO₂), 1537, 1638 (C=N), 3354 (NH). [α]²⁵_D = +29.2 (c = 0.51, CHCl₃). HR-EI-MS: 516.1750 ($C_{29}H_{29}N_2O_3PS^+$; calc. 516.1636).

5. Representative Procedure for the Allylic Alkylation Reaction. A Schlenk tube was charged with $(CuOTf)_2 \cdot C_6H_6$ (4.3 mg, 0.0085 mmol) and the ligand **11ce** (10 mg, 0.018 mmol) under Ar. After addition of THF (0.5 ml), the mixture was stirred at r.t. for 45 min. A soln. of diethyl 3-phenylprop-2-enyl phosphate (**12**; 47 mg, 0.174 mmol) in THF (0.5 ml) was then added. After 15 min, the mixture was cooled to -78° , and 1M Et₂Zn in toluene (0.522 mmol) was added dropwise. The mixture was allowed to warm to -55° , stirred for 20 h, and quenched by addition of a sat. NH₄Cl soln. A sample of the org. layer was analyzed by GC for the determination of conversion and ee. The org. layer was evaporated and the residue purified by FC (hexane): **13**/**14**, inseparable mixture. GC (*Megadex DACTBSβ*, 25 m, film 0.25 µm; H₂ (70 kPa); injector 200°, detector 200°, oven temp. 50°; 1°/min to 70°, then 4°/min to 200°): t_R (**13**) 13.9 ((3R) enantiomer) and 14.1 ((3S) enantiomer), t_R (**14**) 27.7, t_R (**12**) 53.5.

REFERENCES

- C. Gennari, H. P. Nestler, U. Piarulli, B. Salom, *Liebigs Ann./Recl.* 1997, 637; K. D. Shimizu, M. L. Snapper, A. H. Hoveyda, *Chem.-Eur. J.* 1998, 4, 1885; B. Jandeleit, D. J. Schaefer, T. S. Powers, H. W. Turner, W. H. Weinberg, *Angew. Chem., Int. Ed.* 1999, 38, 2494; M. T. Reetz, *Angew. Chem., Int. Ed.* 2001, 40, 284; S. Dahmen, S. Bräse, *Synthesis* 2001, 1431; M. T. Reetz, *Angew. Chem., Int. Ed.* 2002, 41, 1335 and ref. cit. therein.
- [2] a) G. Liu, J. A. Ellman, J. Org. Chem. 1995, 60, 7712; b) K. D. Shimizu, B. M. Cole, C. A. Krueger, K. W. Kuntz, M. L. Snapper, A. H. Hoveyda, Angew. Chem., Int. Ed. 1997, 36, 1704; c) C. Gennari, S. Ceccarelli, U. Piarulli, C. A. G. N. Montalbetti, R. F. W. Jackson, J. Org. Chem. 1998, 63, 5312; d) A. M. Porte, J.Reibenspies, K. Burgess, J. Am. Chem. Soc. 1998, 120, 9180; e) S. R. Gilbertson, X. Wang, Tetrahedron 1999, 55, 11609; f) B. M. Francis, E. N. Jacobsen, Angew. Chem., Int. Ed. 1999, 38, 937; g) M. T. Reetz, M. H. Becker, H. W. Klein, D. Stöckigt, Angew. Chem., Int. Ed. 1999, 38, 1758; h) K. Ding, A. Ishii, K. Mikami, Angew. Chem., Int. Ed. 1999, 38, 497; i) M. S. Sigman, P. Vachal, E. N. Jacobsen, Angew. Chem., Int. Ed. 2000, 39, 1279; j) M. T. Reetz, K. E. Jaeger, Chem.–Eur. J. 2000, 6, 407; l) J. Tian, G. W. Coates, Angew. Chem., Int. Ed. 2000, 39, 3891; k) M. T. Reetz, K. E. Jaeger, Chem.–Eur. J. 2000, 6, 407; l) J. Tian, G. W. Coates, Angew. Chem., Int. Ed. 2000, 39, 3626.

- [3] a) I. Chataigner, C. Gennari, U. Piarulli, S. Ceccarelli, Angew. Chem., Int. Ed. 2000, 39, 916; b) S. Ongeri,
 U. Piarulli, R. F. W. Jackson, C. Gennari, Eur. J. Org. Chem. 2001, 803; c) I. Chataigner, C. Gennari, S. Ongeri, U. Piarulli, S. Ceccarelli, Chem.-Eur. J. 2001, 7, 2628.
- [4] a) M. Gude, U. Piarulli, D. Potenza, B. Salom, C. Gennari, *Tetrahedron Lett.* 1996, 37, 8589; b) C. Gennari,
 M. Gude, D. Potenza, U. Piarulli, *Chem.-Eur. J.* 1998, 4, 1924.
- [5] J. R. Porter, J. F. Traverse, A. H. Hoveyda, M. L. Snapper, J. Am. Chem. Soc. 2001, 123, 984.
- [6] S. J. Degrado, H. Mizutani, A. H. Hoveyda, J. Am. Chem. Soc. 2001, 123, 755.
- [7] D. Wernic, J. DiMaio, J. Adams, J. Org. Chem. 1989, 54, 4224.
- [8] G. C. Lloyd-Jones, S. C. Stephen, Chem.-Eur. J. 1998, 4, 2539.
- [9] a) F. Dübner, P. Knochel, Angew. Chem., Int. Ed. 1999, 38, 379; b) C. A. Luchaco-Cullis, H. Mizutani, K. E. Murphy, A. H. Hoveyda, Angew. Chem., Int. Ed. 2001, 40, 1456; c) H. Malda, A. W. van Zijl, L. A. Arnold, B. L. Feringa, Org. Lett. 2001, 3, 1169.
- [10] G. J. Meuzelaar, A. S. E. Karlström, M. van Klaveren, E. S. M. Persson, A. del Villar, G. van Koten, J. E. Bäckvall, *Tetrahedron* 2000, *56*, 2895; A. S. E. Karlström, F. F. Huerta, G. J. Meuzelaar, J. E. Bäckvall, *Synlett* 2001, 923; A. Alexakis, C. Malan, L. Lea, C. Benhaim, X. Fournoux, *Synlett* 2001, 927.
- [11] T. Hayashi, T. Hagihara, Y. Katsuro, M. Kumada, Bull. Chem. Soc. Jpn. 1983, 56, 363.
- [12] W. C. Still, M. Kahn, A. Mitra, J. Org. Chem. 1978, 43, 2923.

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